Neutrons and Soft Matter: From Soap Bubbles to Cell Membranes

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SOFT MATTER

"Molecular systems giving a strong response to very weak command signal"



"founding father of soft matter"

Condensed matter: states are easily deformed by small external fields, including thermal stresses and thermal fluctuations.

Relevant energy scale comparable with room temperature thermal energy.

Structures in the size range of **nanometres to a few micrometres**. Complex fluids : including colloids, polymers, surfactants, foams, gels, liquid crystals, granular and biological materials.





Soft matter plays an important role in nearly every aspect of our daily life and soft matter research is a driving force in a broad range of innovation fields.





method hand wash

Discoverskips.



Detto

Manual

package contain

Products where surfactant is a secondary component in the material or the production.

GTX

E.

ПX

GOILD

ពា

Materials with very different functions sharing common structural features



Soap bubbles and cell membranes are formed by **amphiphilic** molecules able to **self-assemble**, a few **nanometer** thick and which **structure** and **dynamics** can be determined by **scattering** techniques

Hydrophobic Effect

Tendency of nonpolar substances to aggregate in aqueous solution and exclude water molecules

It explains the separation of a mixture of oil and water into its two components, and the beading of water on nonpolar surfaces.



entropic effect originating from the disruption of highly dynamic hydrogen bonds between molecules of liquid water by the nonpolar solute

At the molecular level, it is important in driving protein folding, formation of **lipid bilayers and micelles**, insertion of membrane proteins into the nonpolar lipid environment and protein-small molecule interactions.

Self-assembling amphiphilic systems:

alcohols, surfactants or soap like molecules and lipids





Surface Tension

The cohesive forces between molecules down into a liquid are shared with all neighboring atoms.

Those on the surface have no neighboring atoms above, and exhibit stronger attractive forces upon their nearest neighbors on the surface.

This enhancement of the intermolecular attractive forces at the surface is called **surface tension**.





ramé-hart instrument co.

Pure water has a surface tension of ~ 72 mN/m, but a monolayer can cause this to drop nearly to zero.









Lung Surfactants:

The normal surface tension in the lungs is 25 mN/m; at the end of the expiration, compressed surfactant phospholipid molecules decrease the surface tension to near-zero levels.

Pulmonary surfactant allow the lung to inflate much more easily, thereby eliminating the work of breathing. It reduces the pressure difference needed to allow the lung to inflate



Premature infants lacking of these surfactants suffer from infant respiratory distress syndrome



From soap to lipid bilayers





Lipids are amphiphiles:

in aqueous environment self-assemble (reduced specific volume ~1 *i.e.* bilayers are formed)

Highly hydrophobic core forms a barrier : protects content of cell
Lipid bilayer participates to exchanges between extracellular fluid and cytoplasm

The structure and organization of the lipid bilayer component of membranes hold the key to understanding the functioning of membranes

Why important?

Total surface of membranes covers an area of ~100 m² in our body

Function of membrane proteins : dependent on membrane composition, lipid-protein interaction, lipid mediated protein-protein interaction

Pharmacological interest : Drug transport through membranes (dependent on physico-chemical membrane properties), anti-microbial peptides

Membranes may play a direct role in **signal transduction**

Diseases associated with changes in lipid composition (heart disease, obesity, diabetes, cancer and neurodegenerative disorders like schizophrenia, Tay-Sachs syndrome, Alzheimer, Parkinson)

Cell adhesion



Na*, K*-ATPasa activity + choiesterol 400 1.5 1.0 0.5 30 Retributed in New Yorking Ions old chain langth (RC Sodiu Intracellular space Drug transport in liposomes

Nanobiotechnology Applications



A new type of biosensor is based on a lipidcoated nanotube

Catalytic biosensors (for example glucose biosensors) Affinity biosensors (antibodies, DNA, peptides and lectins)

Current problems include: non-specific binding, reproducibility

Biofunctional Coatings: Artificial Organs and Implanted Medical Devices

Why Neutron Scattering?

Probe relevant length (**Å to** µm) and time-scales (ps to hr) Sensitive to light elements Buried systems and complex sample environment Possibility of **isotopic labelling**

Non-destructive





Spin-dependent scattering lengths

Calculation of the scattering length density

neutrons deflected from hydrogen are 180° out of phase relative to those deflected by the other elements





SCATTERING



time scales from about 0.1 ps to almost 1 μs





http://www.rheinstaedter.de/maikel/

Collective excitations

Inelastic, Backscattering, Spin-Echo



Protein Crystallography - neutrons see hydrogens crystal deuteriation allowes smaller samples and higher resolution data

down to 2.0 Å resolution times ~ days

LADI-III



Diffraction

2-D info ~Å to ~10nm times ~ min









Membrane proteins structural characterisation by using nanodiscs





with **contrast variation** in SANS it is possible to highlight the interesting parts of the system





SANS signal expected from membrane proteins



 SAXS from Sensoryrhodopsin-II

Courtesy Selma Maric



Specular $\theta_i = \theta_f$

•Thickness of layers at interfaces

Roughness/interdiffusion

•Composition in the direction normal to the interface



Reflectivity measurements:



Specular $\theta_i = \theta_f$

•Thickness of layers at interfaces

Roughness/interdiffusion

•Composition in the direction normal to the interface



Reflectivity measurements:



In-plane features (height fluctuations, domains, holes ...) can be probed by off-specular measurements: for thin films synchrotron radiation is more suitable For both kinds of radiation the refractive index is a function of the scattering length density and wavelength.



As with light, total reflection may occur when neutrons pass from a medium of higher refractive index to one of lower refractive index.

Two well known examples of light interference from thin films

Oxidized oil film on water





Drainage of soap film. Note the absence of reflection in the top half of the film in 6.

Optical Demonstrations

Reflection from a thin film Newton's Rings

Courtesy of R. K. Thomas http://rkt.chem.ox.ac.uk/techniques/nrmain.html



1.E+00

Scattering length density profile extracted from data analysis



$$R(\vartheta,\lambda) = \frac{I_{out}(\vartheta,\lambda)}{I_{in}(\lambda)}$$

$$q = \frac{4\pi}{\lambda} \sin \theta$$



$$N_b = \frac{\sum_i n_i b_i}{V}$$

Scattering length density related to layer composition



Reflected intensity depends on structure and composition at the interface

Born Approximation

 $q>>q_{c}$ $q_{c}=\sqrt{16\pi N_{b}}$

Ignored double scattering processes because these are usually very weak

$$R(q) = \frac{16\pi^2}{q^4} \left| N_b'(q) \right|^2$$

$$N_{b}'(q) = \int_{-\infty}^{+\infty} \exp(iqz) \frac{dN_{b}}{dz} dz$$

Schema of sample holder



Data modelling

Thin Film – Neutron Reflection Contrast

Courtesy of R. K. Thomas http://rkt.chem.ox.ac.uk/techniques/nrmain.html



Understanding Intracellular Cholesterol Transport

Garg et al. 2011

In-vivo

In-vitro



Aim: validate results from in-situ Transfer by Time-Resolved SANS



1) Does not require vesicles isolation (in situ technique)

2) No need of fluorescent or tag cholesterol

- 3) Access early stage of transfer
- 4) Accurate control on lipid membrane composition and structure
- 5) Can be applied to anything (just require deuterated materials)

Cholesterol Exchange

Lipid Exchange

Time-Resolved SANS approach (Garg et al., NIST, Biophys. J., 2011)



Kinetics

 $\frac{dC_{in1}}{dt} = -K_f C_{in1} + K_f C_{out1}$

$$\frac{dC_{out1}}{dt} = K_f C_{in1} - K_f C_{out1} - K_{ex} C_{out1} + K_{ex} C_{out2}$$

Scattering

 $I \sim \phi_{Vesl} V_1 (SLD_{vesl} - SLD_{solvent})^2 FF$ $+ \phi_{Ves2} V_2 (SLD_{ves2} - SLD_{solvent})^2 FF$

$$I \sim \frac{\Phi_{Ves}}{2} \left\{ \left(\left(\phi_{chol1}^{vol} \right)^2 + \left(\phi_{chol2}^{vol} \right)^2 \right) SLD_{Chol} - SLD_{solvent} \right]^2 \right\}$$



Cholesterol's transfer in POPC vesicles





Total cholesterol exchange!!

Half life for exchange: ~100min Half life for flipping: ~250min (surprisingly slow)

Comparison with MD simulations and literature



Asymmetric deposition 12°C 40 mN/m lateral pressure d83-DSPC by LB (inner) DMPC by LS (outer)





Thermal cycle for NR measurements $14^{\circ}C \rightarrow 30^{\circ}C \rightarrow 60^{\circ}C \rightarrow 30^{\circ}C \rightarrow 14^{\circ}C$

T_m T_m d83-DSPC DMPC 50.5°C 23.5°C





30°C→60°C ... also DSPC becomes fluid



10⁻² 10⁻² 10⁻³ 10⁻³ 10⁻⁴ 10⁻⁵ 0.05 0.10 Q (Å⁻¹)

Slight decrease of the total thickness

Decrease of internal contrast



Gerelli Y., et al., Langmuir 2012(28), 15922







intermediate points characterised in



B

Gerelli Y., et al., Langmuir 2013(29), 12762





$$\alpha(t, T) = \alpha(0) e^{-t/\tau(T)} \qquad E_a^{tx} \equiv R \left[\frac{\partial \ln \tau(T)}{\partial (1/T)} \right]$$

Technique	E _a
I:I LUV–LUV Exchange	85 ± 2
LUV–Bilayer Exchange	81 ± 7

E 9 0.0029 0.0030 0.0031 1/T (K⁻¹)

^a Nakano, M. et al. Phys. Rev. Lett. 2007, 98, 238101

Gerelli Y., et al., Langmuir 2013(29), 12762

For pure DMPC :

Preliminary indications of a disagreement with SANS data, flip-flop seems to be faster than the exchange and therefore NOT visible.

Is this due to <u>planar geometry and coupling to the</u> <u>substrate</u> in the NR case?

or Is this due to the <u>different experimental conditions</u> (h/d-lipids relative concentrations)?

Essential to select well the sample conditions

Neutron reflectometry and deuteriation to probe density profiles of proteins adsorbed onto polymer brushes



polymer-functionalized surface

Emanuel Schneck, Audrey Schollier, Avi Halperin, Michele Sferrazza, Michael Haertlein, Martine Moulin

Density Profiles of Proteins in Polymer brushes



substrate



implants

catheters/stents



biocompatible surface functionalization

"brush failure" via protein adsorption

modes of protein adsorption: primary, secondary, ternary

bilayer structural characterization for "rational design" of protein resistant functionalization (role of grafting density and polymer length)



Sample Preparation

silicon

1288523

83885

Preparation steps

- Planar silicon substrates
- Hydrophobic functionalization
- Brushes at air/water interface (Langmuir trough) of PS-PEG diblock copolymers or PE-PEG lipid anchored polymers

Resulting brush

- defined grafting density, σ
- defined polymer length, N
- hydrophilic/hydrophobic grafting surface

Scattering length density profile



SLD profile $\varrho(z)$ gives the profile of the organic compounds;

NOT UNIQUE:

 Unambiguous result can be obtained by step-wise build-up of sample architecture; use of contrast variation

SIMULTANEOUS ANALYSIS of 8 reflectivity curves/ sample (4 water contrasts before and after protein adsorption)

Schneck, Schollier et al., Langmuir 2013



Layers below grafting surface

Data Analysis

 slabs with adjustable thickness, dry SLD, water content, interface roughness



PEG brush

parabola (SCF theory) with adjustable brush length and density

After protein adsorption

protein distribution that allows for primary, secondary, ternary adsorption (rough slabs+Gaussians)

SLD of PEG and protein fixed

Schneck, Schollier et al., Langmuir 2013

 dependence of protein SLD on water contrast (H/D exchange) taken into account

Data Analysis



for each parameter set:

- compute SLD profiles corresponding to all measurement conditions;
- discretized into 1Å slices;
- compute corresponding reflectivity curves (dynamical treatment: Fresnel reflection coefficients, Parrat formalism)

parameters are varied to achieve best agreement between measured and modelled reflectivity curves

Bare BrushesResults consistent with SCF theoryPEG 114<N<770 up to σ ~ 2x10¹⁷m⁻¹ (5nm² per chain)

* parabolic brush model gives density, Φ_0 and length, H_0



Adsorption of <u>deuterated</u> myoglobin to PEG brushes grafted on hydrophobic polystyrene surfaces





 Significant adsorption for all brush parameters

only primary adsorption

Adsorption of <u>deuterated</u> myoglobin to PEG brushes grafted on hydrophobic polystyrene surfaces

- inner-layer: protein amount decreases with grafting density
- anchoring points obstacles adsorption
- outer protein layer depends on overall PEG amount and protein-protein interactions are altered by the presence of PEG
- Information only accessible with neutron reflection combined with protein perdeuteration



Specific adsorption: PEG antibodies

- Classically PEG purely repellent, in fact it is antigenic
- PEG antibodies produces in animals (0.1% 25% in humans)
- Implications on brush functioning failure?
- IgG AB bind specifically to end segments of PEG







substrate

Brushes grafted to hydrophilic phospholipid surface to prevent primary adsorption

Specific adsorption: PEG antibodies

Neutron reflectometry measurements



Brushes grafted to hydrophilic phospholipid surface to prevent primary adsorption

Specific adsorption: PEG antibodies

- Antibodies adsorbs at brush periphery
- No primary adsorption
- Amount increases with grafting density
 Saturation molecular crowding

Antibodies become the dominant surface: Brush no more functional foreign-body reaction





Schneck et al., in preparation

 Many open questions regarding protein adsorption to polymer brushes

- Neutron reflectometry (coupled to protein deuteriation) promising approach
- Detailed structural insight

 Unique tool to investigate the structure of biological interfaces and interfaces relevant for biotechnological applications

Conclusions

- Neutron scattering remains an essential tool for the study of structure at the nanometer level of soft self-assembled systems.
- Complementary to x-ray and synchrotron radiation, advantages include high penetration, sensitivity to light elements (H, C, O, N, ...) and isotopic labelling/contrast variation.
- * Possibility to work in real (physiological) conditions
- * Possibility for in-situ studies of systems under deformation.
- * Need optimised sample preparation
- Perspectives in biology are very numerous.

Thank you for your attention

