Neutron production

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What is a neutron?

1. a subatomic particle

2. a matter wave





Neutrons are everywhere



Bound neutrons are everywhere Others - Nitrogen 3% 10% Hydrogen -Carbon -18% 65% Carbon-13 Carbon-12 - Oxygen 98.9% 1.1% 6 protons 6 protons 7 neutrons 6 neutrons 45% neutrons



The Neutron's Circle of Life

- 1. How neutrons are born
- 2. How neutrons are conformed to use
- 3. How neutrons die
- 4. What neutrons are good for (except neutron scattering and nuclear spectroscopy)

1. Alpha-induced reactions: ${}^{9}Be(\alpha,n){}^{12}C$ +5.7 MeV

| C 11 | C 12 | C 13 | |
|--|--|--|--|
| 20.38 m | 98.93 | 1.07 | |
| β ⁺ 1.0 no γ | σ 0.003 5 | σ 0.0014 | |
| B 10 | B 11 | B 12 | |
| 19.9 | 80.1 | 20.20 ms | |
| σ 0.3 σ _{n.α} 3840 σ _{n.p} 0.007 | σ 0.005 | β 13.4 γ 4439 βα 0.2 | |
| Be 9 | Be 10 | Be 11 | |
| 100 | 1.387·10 ⁶ a | 13.8 s | |
| σ 0.0078 | β [−] 0.6 noγ σ < 0.001 | β ⁻ 11.5 γ 2125, 6791 βα 0.77, 0.29 | |

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- 2. Deuteron fusion: d(d,n)³He +3.3 MeV, t(d,n)⁴He +17.6 MeV

| He | He 3 | He 4 | | |
|-------------------------|------------------------------------|-------------------------------------|--|--|
| 4.002602 | 0.000134 | 99.999866 | | |
| σ _{abs} < 0.05 | σ 0.00005 σ _{n.p} 5330 | | | |
| H 1 | H 2 | H 3 | | |
| 99.9885 | 0.0115 | 12.312 a | | |
| σ 0.332 | σ 0.00051 | β 0.0185743 σ < 6E-6 | | |

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| Kr 86 17.279 | Kr 87 76.3 m | Kr 88 2.84 h β ⁻ 0.5, 2.9 | | |
|--------------------------------------|--|---|--|--|
| σ 0.003 | β 3.5, 3.9 γ 403, 2555 845 | 2392, 196 2196, 835 1530 | | |
| Br 85 2.87 m | Br 86 55.1 s | Br 87 55.7 s | | |
| β 2.5 γ 802, 925 m | β 3.3, 7.6 γ 1565, 2751 | p ⁻ 6.8. + 1420, 1476, 1578 532, 2006 552, 0.05 | | |

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High energy nuclear reactions





T. Enqvist et al., Nucl. Phys. A 686 (2001) 481.

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A nuclear chain reaction



A single-pulse neutron source



Uncontrolled chain reaction of fast-neutron induced fission

≈25 kg of 93%²³⁵U







²³⁵U(n,f) cross-section as function of energy



 \approx 0.6% of fission neutrons are beta-delayed by 12 s on average \Rightarrow slows down reactor kinetics ($\Delta k = 0.001$) from ≈ 0.05 s to ≈ 80 s \Rightarrow essential for reliable control of reactor power



Research reactor

Components of a nuclear reactor

- 1. Fuel
- 2. Moderator
- 3. Control rods
- 4. Coolant
- 5. Pressure vessel
- 6. Containment
- Steam generator (for power plants) or experimental facilities (for research reactors)

Moderator

elastic collisions with light atoms (mass A): average energy loss $E_{n+1} - E_n = 2 E_n A/(A+1)^2$

 $ln(E_n) - ln(E_{n+1}) = \xi = 1 - (A-1)^2/(2A) * ln[(A+1)/(A-1)]$

| Moderating power: | $\xi \Sigma_{scatter}$ | |
|--------------------------------|------------------------|--|
| Moderating ratio: | | $\xi \Sigma_{\text{scatter}} / \Sigma_{\text{abs.}}$ |
| Light water (H ₂ O) | 1.28 | 58 |
| Heavy water (D ₂ O) | 0.18 | 21000 |
| Beryllium (Be) | 0.16 | 130 |
| Graphite (C) | 0.064 | 200 |
| Polyethylene $(CH_2)_x$ | 3.26 | 122 |

The first nuclear reactor on Earth



Choice of coolant

coolant = moderator \Rightarrow passive regulation \Rightarrow intrinsic safety

RBMK:

graphite moderator water cooling ⇒ positive void coefficient !



RHF fuel element



8 December 1987: Intermediate-Range Nuclear Forces Treaty











Some comments on recent events...

Reactor fuel elements = 1st barrier





Typical boiling-water reactor









Decay heat can be passively cooled by natural convection!

Secondary reactions



Safety features of the ILL reactor



Safety features of Generation 3+ reactors (EPR)



Power reactor

- heat used to produce electricity
- neutrons just to maintain heat not used chain reaction
- needs high power, high temperature and high pressure for good thermal efficiency
- BWR: 75 bar, 285°C
- PWR: 155 bar, 315°C
- 25 cm thick steel pressure vessel \Rightarrow defines lifetime (40..60 y)

Research reactor

- neutrons used for applications
- operates at lower power, low temperature (ILL 30-48°C) and low pressure (<14 bar)
- · vessel and all inserts made from pure Al-alloy
- modular and exchangeable \Rightarrow no finite lifetime

ILL: Replacement of the reactor vessel 1990-94



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The risk profile of power versus research reactors



Spallation neutron source versus reactor

Advantage for reactor

- higher time-averaged flux (for high-flux reactor)
- flux very constant over time
- larger irradiation volume possible (multiple fuel elements)
- much lower electricity bill

Advantage for spallation neutron source

- pulsed operation much easier
- much higher peak flux for TOF applications
- does not carry "reactor" in its name

Urban legend:

"Reactors are dirty, spallation neutron sources are clean."

Back to physics:

the neutron as a tool to study fission

LOHENGRIN Setup



¹¹Li production in thermal neutron induced fission?



Detection of rare ternary particles A/q = 11/3 E/q = 5 MeV Exotic ??? ∆E (channels) 10¹³ per s produced worldwide in nuclear B power plants! Be 2.10⁻¹⁰ per fission E_{total} (channels)

Neutron (particle) physics:

the neutron as an object

Neutron Decay

• clean semi-leptonic decay

 $n \to p + e^- + \overline{\nu}_e$

• clear theoretical understanding:

only 3 free parameters in Standard Model

- ratio of coupling constants
- and its phase
- Quark mixing from the Cabibbo-Kobayashi-Maskawa matrix





n(udd)

$$\tau^{-1} \propto \left| V_{ud} \right|^2 \left(1 + 3\lambda^2 \right)$$

$$\lambda = \frac{g_A}{g_V}$$

 V_{ud}

Why is neutron-decay important?

neutron-decay provides key input for many disciplines: astrophysics, cosmology and particle physics



From cold to ultracold neutrons

Ultracold neutrons



Total reflection under any angle on suitable materials like Be, Ni, C \Rightarrow possibility to store for long time



The UCN facility PF2 at ILL

Measurements of the neutron lifetime $\tau_{\!n}$



A "typical" UCN storage experiment at ILL – MamBo I



Glass w alls: H=0.3 m, W=0.4 m L=0.5m ... 0.01 m (surface A and volume V sizeable)

$$\frac{1}{\tau_m} = \frac{1}{\tau_\beta} + \frac{1}{\tau_{\text{wall}}} + \dots$$

$$\mathfrak{r}_{wall} \rightarrow$$
 number of wall collisions
i.e. mean free path λ

Measure storage lifetime τ_{st} for different volume to surface ratios V/A and extrapolate for $V \rightarrow \infty$

$$\frac{1}{\tau_{\text{wall}}} \to 0$$

"GRAVIT RAP" Neutron Lifetime Experiment at the PF2/MAM beam position in the ILL



Extrapolation to n-lifetime



Magnetic confinement

- For μ_n = -60.3 neV/T, a 2T field generates a 120 neV barrier.
- Force due to field gradient, $F = -\mu (dB/dz)$, repels only one spin state.
- Use permanent magnets.







Scales of temperature and energy in neutron physics

Neutrons as a tool for medicine



Cancer and efficiency of treatments

| At time of diagnosis | Primary tumor | With metastases | Total |
|--|------------------|--------------------|-------|
| Diagnosed | 58% | 42% | 100% |
| Cured by: | | | |
| Surgery | 22% | | |
| Radiation therapy | 12% | | |
| Surgery+radiation therapy | 6% | | |
| All other treatments and combinations incl. chemotherapy | | 5% | |
| Fraction cured | 69% | 12% | 45% |

Over one million deaths per year from cancer in EU.

 \Rightarrow improve early diagnosis

 \Rightarrow improve systemic treatments

Immunology approach





Nuclear medicine and medical physics





Lymphoma therapy: RITUXIMAB+¹⁷⁷Lu E.B., 1941 (m): UPN 6



Alternative production route to ¹⁷⁷Lu

| Ta 175 10.5 h | Ta 176 8.1 h | Ta 177 56.6 h | Ta 178 9.25 m 2.45 h | Ta 179 665 d | Ta 180 0.012 | Ta 181 99.988 |
|---|---|--|--|---|--|---|
| ε γ 207; 349; 267; 82; 126; 1793 | ε β ⁺ γ 1159; 88; 1225 | ε β ⁺ γ 113; 208 9 | β ⁺ 0.9 γ 93; 1351; ε 1341 γ 332 g m ₁ | ε no γ g σ 930 | σ~560 g | σ 0.012 + 20 σ _{n, α} <10 ^ε |
| Hf 174 0.16 2.0 · 10 ¹⁵ a α 2.50 σ 600 | Hf 175 70.0 d | Hf 176 5.26 or 23 | Hf 177 51 m 1.1 s 18.60 ¹ y ¹ y 208; 0:10 ⁻⁷ 295; 229; +1 327 379 + 375 | Hf 178 31 a 4.0 s 27.28 hy hy 574; 426; 495; 326; 7? 217 213; + 54 or 45 89 + 32 | Hf 179 25 d 18.7 s 13.62 ¹ y ^{454;} 363; 123; ¹ y214 146 ¹ y 46 | Hf 180 5.5 h 35.08 1y 332; 443; 215; 57 β ⁻ m ⁻¹ <1.3 - 1 -1 |
| Lu 173 1.37 a ^ε γ 272; 79; 101 e ⁻ | Lu 174 142 d 3.31 a hy45; 67 ε σ ⁻ ;ε β ⁺ γ (1992; γ 1242; 76 | Lu 175 97.41 0 ^{16 + 8} | Lu 176 2.59 3.68 h 38.10 ¹⁰ a β ⁻¹ 12; 13;ε γ 88 φ ⁻ 02.83 902;88 φ ⁻² 24.2100 | $\begin{array}{c c} Lu & 177 \\ \hline 160.1 d & 6.71 d \\ \beta^{-0.2} & \beta^{-0.5} & \\ \gamma 414; & \\ 319; 122 & 113 \\ m; & g \\ \sigma .3.2 & \sigma 1000 \end{array}$ | Lu 178 22.7 m 28.4 m β ⁻ 2.0 γ33: γ332 m ₁ 1269; g | Lu 179 4.6 h β ⁻ 1.4 γ214 9 |
| Yb 172 21.83 | Yb 173 16.13 | Yb 174 31.83 | Yb 175 4.2 d | Yb 176 | 15 177 6.5 s 1.9 h | Yb 178 74 m |
| σ~1.3 σ _{n, α} <1E-6 | σ16 σ _{n. α} <1E-6 | σ 63 σ _{n, α} <0.00002 | β ⁺⁻ 0.5 γ 396; 283; 114 | ly 293 390; 190; <mark>073.1</mark> 96 | ly 104; 228 e g | β 0.6 γ 391; 348; 9 |
| Tm 171 1.92 a | Tm 172 63.6 h | • Free of long-lived isomer | | | Tm 177 85 s | |
| β ⁻ 0.1 γ (67); e ⁻ σ~160 | 0.1 β ^{-1,8;1,9,} • Non-carrier-added quality 7/2; 1034; β • Non-carrier-added quality 1387; 1530; β • "Needs" high-flux reactor | | | | β γ 105; 518 g; m | |



Acknowledgements

Thanks for transparencies from: Hartmut Abele Roger Brissot Bruno Desbriere Peter Geltenbort Bastian Maerkisch Valery Nesvishevsky Anatoli Serebrov Oliver Zimmer